Supporting Information for

Selective Hydration of Electron-Rich Aryl-Alkynes by a Schrock-type

Molybdenum Alkylidene Catalyst

by

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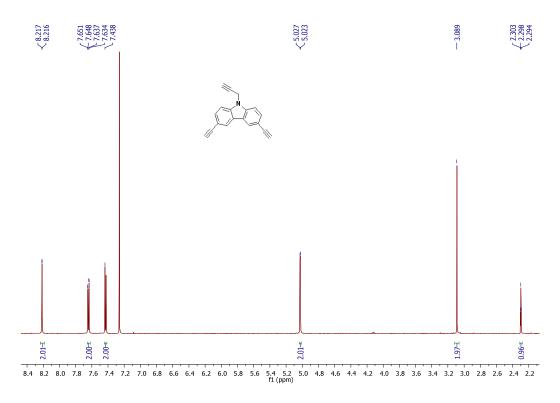


Figure S1. ¹H-NMR (600 MHz, CDCl₃) spectrum of compound **3**

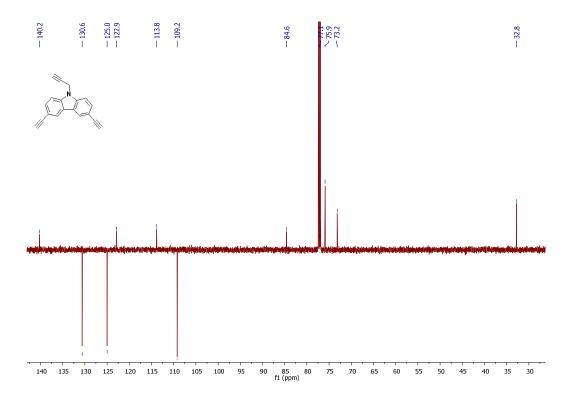


Figure S2. ¹³C-NMR (100 MHz, CDCl₃) spectrum of compound **3**

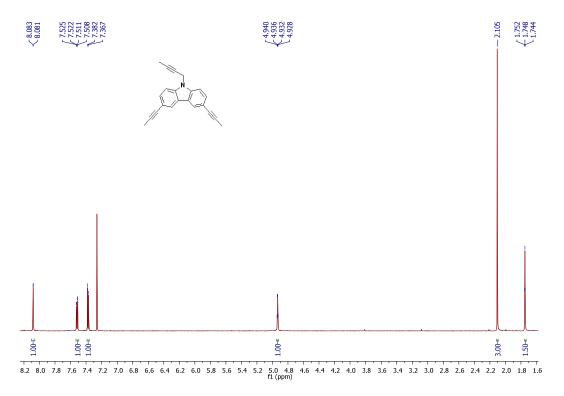


Figure S3. ¹H-NMR (600 MHz, CDCl₃) spectrum of compound 4

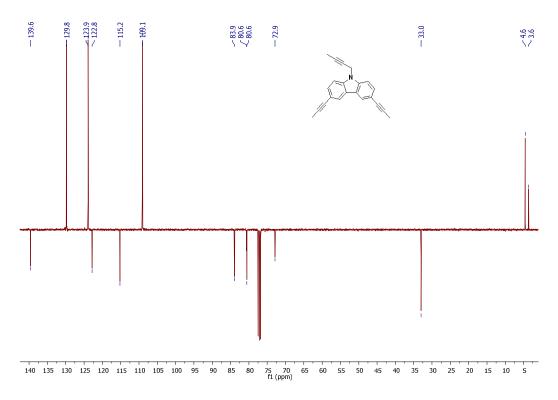


Figure S4. ¹³C NMR (150 MHz, CDCl₃) spectrum of compound **4**

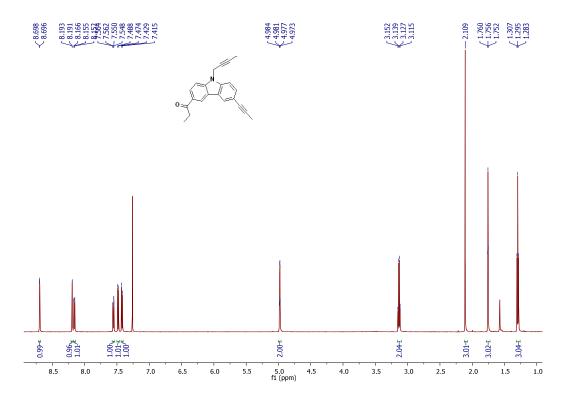


Figure S5. ¹H-NMR (600 MHz, CDCl₃) spectrum of compound 5

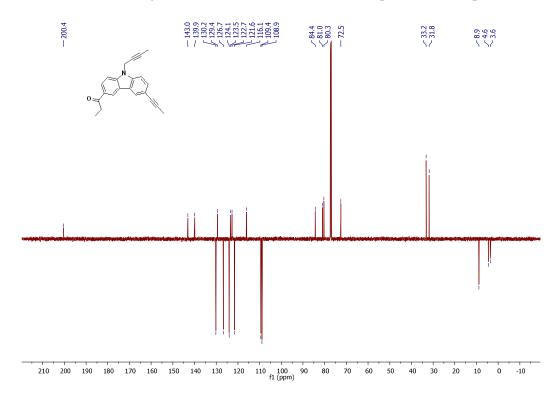


Figure S6. ¹³C NMR (150 MHz, CDCl₃) spectrum of compound **5**

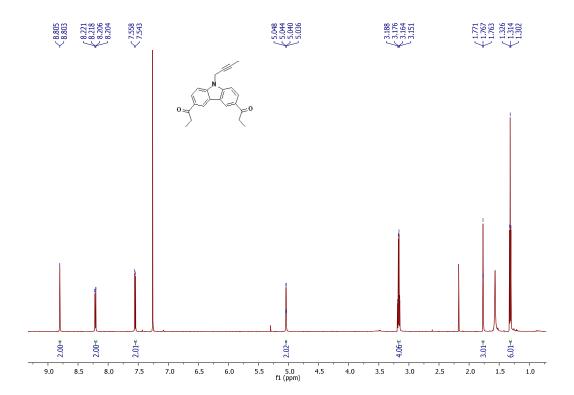


Figure S7. ¹H NMR (600 MHz, CDCl₃) spectrum of compound **6**

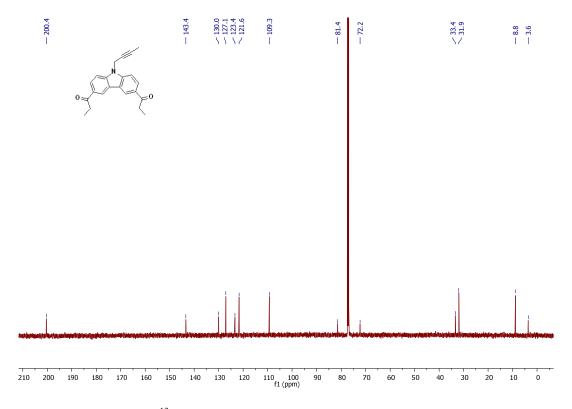


Figure S8. ¹³C NMR (150 MHz, CDCl₃) spectrum of compound **6**

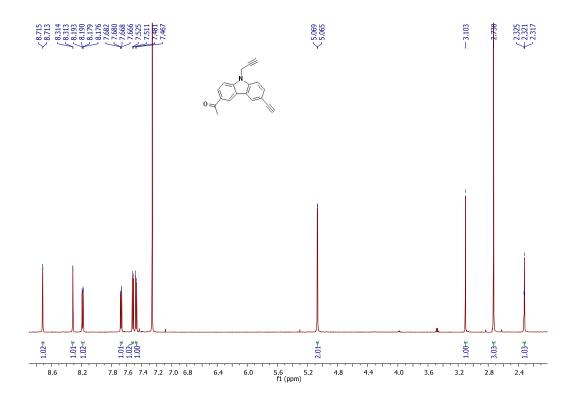


Figure S9. ¹H NMR (600 MHz, CDCl₃) spectrum of compound 7

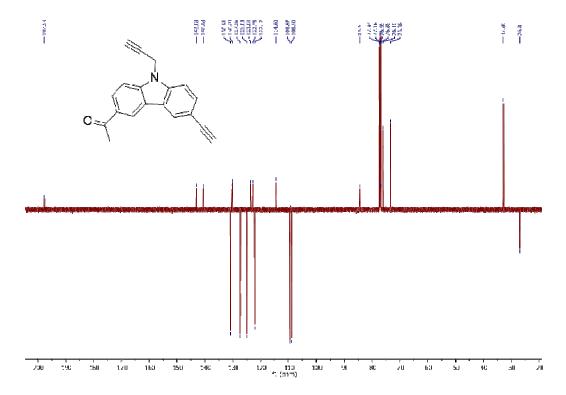


Figure S10. ¹³C NMR (150 MHz, CDCl₃) spectrum of compound **7**

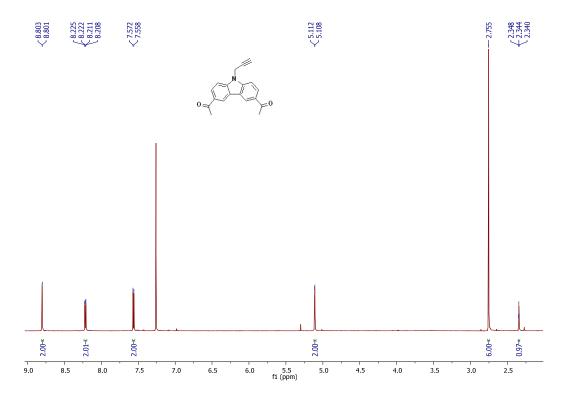


Figure S11. ¹H NMR (600 MHz, CDCl₃) spectrum of compound **8**

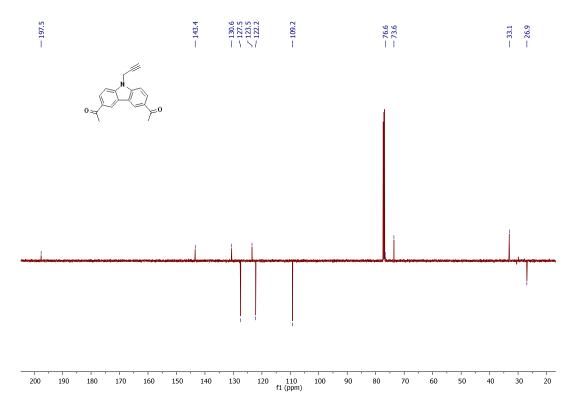


Figure S12. ¹³C NMR (150 MHz, CDCl₃) spectrum of compound **8**

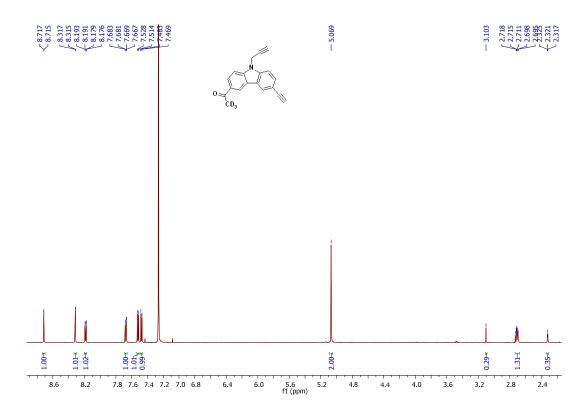


Figure S13. ¹H NMR (600 MHz, CDCl₃) spectrum of compound 9

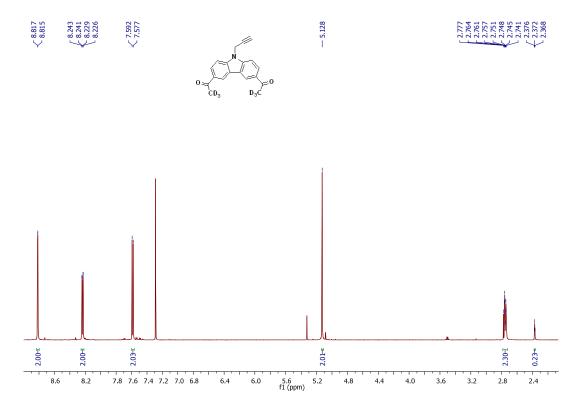


Figure S14. ¹H NMR (600 MHz, CDCl₃) spectrum of compound **10**

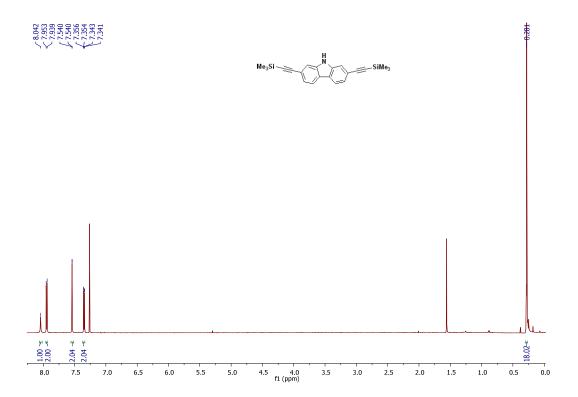


Figure S15. ¹H NMR (600 MHz, CDCl₃) spectrum of compound **11**

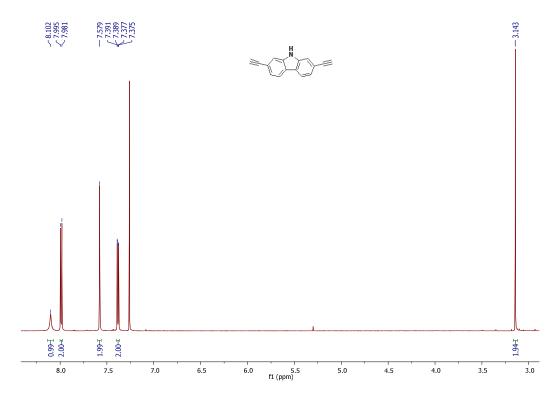


Figure S16. ¹H NMR (600 MHz, CDCl₃) spectrum of compound **12**

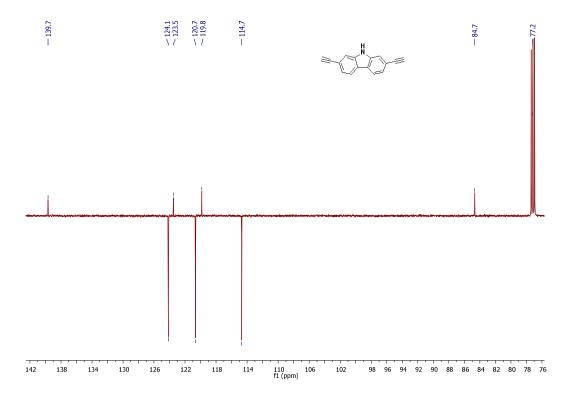


Figure S17. 13 C NMR (150 MHz, CDCl₃) spectrum of compound 12

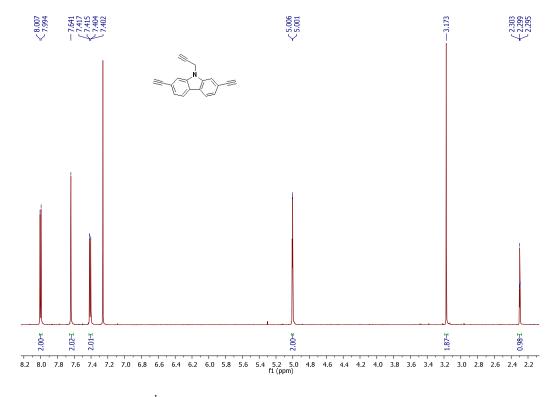


Figure S18. ¹H NMR (600 MHz, CDCl₃) spectrum of compound 13

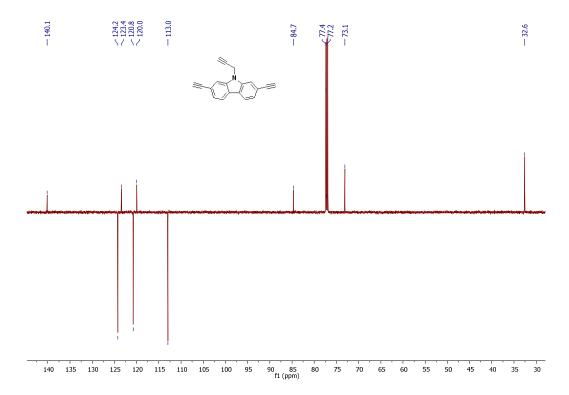


Figure S19. ¹³C NMR (150 MHz, CDCl₃) spectrum of compound **13**

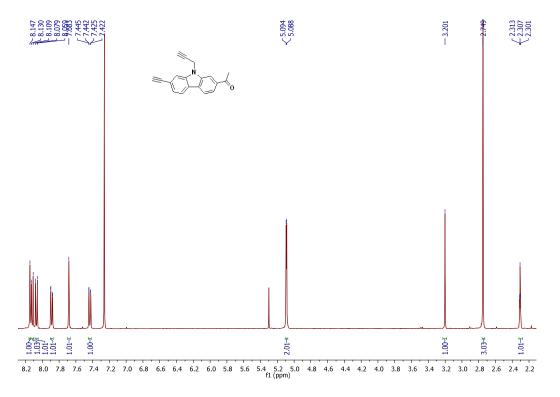


Figure S20. ¹H NMR (600 MHz, CDCl₃) spectrum of compound **14**

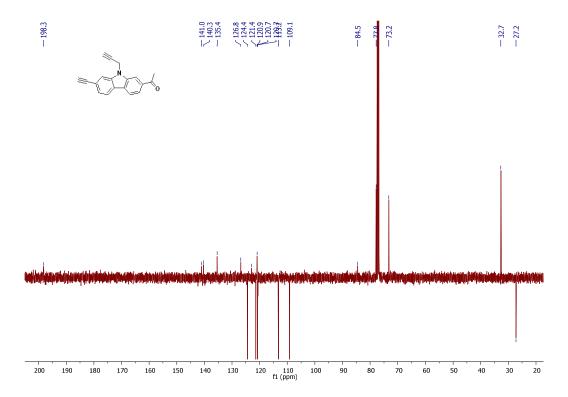


Figure S21. ¹³C NMR (150 MHz, CDCl₃) spectrum of compound **14**

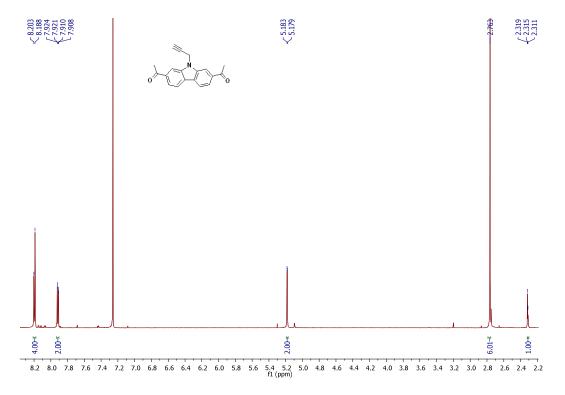


Figure S22. ¹H NMR (600 MHz, CDCl₃) spectrum of compound **15**

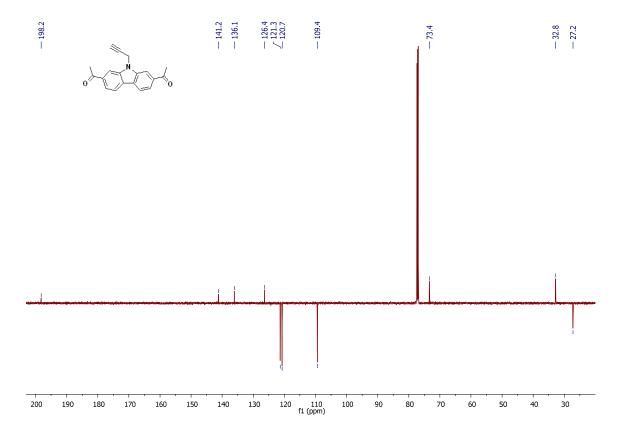


Figure S23. ¹H NMR (150 MHz, CDCl₃) spectrum of compound **15**

Table S1. Crystal data and structure refinement for mono-ketone 7

Identification code shelx

Empirical formula $C_{19}H_{13}NO$ Formula weight 271.30Temperature 294(2) KWavelength 0.71073 ÅCrystal system Monoclinic

Space group P 21/c

Unit cell dimensions a = 13.0200(18) Å $a = 90^{\circ}$.

b = 8.4402(12) Å $b = 107.069(2)^{\circ}$.

c = 13.3130(19) Å $g = 90^{\circ}$.

Volume $1398.5(3) \text{ Å}^3$

Z 4

Density (calculated) 1.289 Mg/m³
Absorption coefficient 0.080 mm⁻¹

F(000) 568

Crystal size $0.390 \times 0.321 \times 0.220 \text{ mm}^3$

Theta range for data collection 1.636 to 24.994°.

Index ranges -15 <= h <= 15, -10 <= k <= 10, -15 <= l <= 15

Reflections collected 13048

Independent reflections 2464 [R(int) = 0.0751]

Completeness to theta = 24.994° 100.0 %

Refinement method Full-matrix least-squares on F²

Data / restraints / parameters 2464 / 0 / 191

Goodness-of-fit on F^2 1.103

Final R indices [I>2sigma(I)] R1 = 0.0811, wR2 = 0.1837 R indices (all data) R1 = 0.1047, wR2 = 0.1957 Largest diff. peak and hole 0.455 and -0.211 e.Å $^{-3}$